

### The Hebrew University of Jerusalem

Syllabus

# General Chemistry For Medicine & Dentistry Stu. - 69134

Last update 31-10-2024

HU Credits: 4.5

<u>Degree/Cycle:</u> 1st degree (Bachelor)

Responsible Department: Chemistry

Academic year: 0

Semester: 1st Semester

<u>Teaching Languages:</u> Hebrew

Campus: E. Safra

Course/Module Coordinator: Moran Frenkel-Pinter

<u>Coordinator Email: moran.fp@mail.huji.ac.il</u>

Coordinator Office Hours: By request

#### **Teaching Staff:**

Dr. Moran Frenkel-Pinter,

Ms. Tehila Matas,

Ms. Roni Cohen,

Mr. Or Cohen

### Course/Module description:

The course will deal with basic concepts in Chemistry: from atoms to molecules to chemical reactions.

### Course/Module aims:

A general introduction to basic concepts in chemistry that would allow to-be physicists to communicate with chemists.

## <u>Learning outcomes - On successful completion of this module, students should be able to:</u>

- 1. Understand basic concepts in Chemistry.
- 2. Describe the atomic structure and the chemical bonding in molecules.
- 3. To identify and classify chemical reactions.

### Attendance requirements(%):

None

Teaching arrangement and method of instruction: Lectures and Exercises

### Course/Module Content:

- 1. Introduction, basic concepts, orders of magnitude, measuring units and units conversion.
- 2. Atomic structure, atomic orbitals, electronic configurations, isotopes.
- 3. Periodic table of elements and its characteristics.
- 4. The chemical bond, Lewis structures, ionic bonding, Covalent bonding.
- 5. Molecules, organic molecules, functional groups, polymers.
- 6. Stoichiometry, The mole concept, Writing and balancing chemical equations, the

limiting factor, concentration in solution.

- 7. Chemical equilibrium, Le Chatelier principle.
- 8. titrations, chemical reactions in solution.
- 9. Oxidation reduction reactions.
- 10. States of matter and intermolecular forces.

### Required Reading:

Please make sure to have a periodic table of the type Sargent Welch

Knowledge in logarithms rules is mandatory

### Additional Reading Material:

General Chemistry / Hill & Pettrucci, 4 th Ed., Pearson Prentice Hall, 2005

Chemical Principles The Quest for Insight/Atkins and Jones, 4th edition, 2008

<u>Grading Scheme:</u>
Written Exam % 85
Mid-terms exams 15 %

Additional information: